## **Syllabus Outline for Chemistry**

<b>Core</b> [80	h]	Teac	Teaching hours	
<b>Topic</b> Sub-topics	1 1.1 1.2 1.3 1.4 1.5	Quantitative Chemistry Mole Concept and Avogadro's Constant Formulas Chemical Equations Mass and gas volume relationships in chemical reactions Solutions	[13] 2 3 1 5 2	
<b>Topic</b> Sub-topics	2 2.1 2.2 2.3	Atomic Theory The Nuclear Atom Physical Properties Chemical Properties	[ <b>4</b> ] 1 1 2	
<b>Topic</b> Sub-topics	3 3.1 3.2 3.3	Periodicity The Periodic Table Physical Properties Chemical Properties	[6] 1 2 3	
<b>Topic</b> Sub-topics	4 4.1 4.2 4.3 4.4 4.5	Bonding Ionic Bond Covalent Bond Intermolecular Forces Metallic Bond Physical Properties	[12.5] 2 6.5 1.5 0.5 2	
<b>Topic</b> Sub-topics	5 5.1 5.2 5.3 5.4	Energetics Exothermic and Endothermic Reactions Calculation of Enthalpy Changes Hess's Law Bond Enthalpies	[8] 1 3 2 2	
<b>Topic</b> Sub-topics	<b>6</b> 6.1 6.2	Kinetics Rates of Reaction Collision Theory	[ <b>5</b> ] 2 3	
<b>Topic</b> Sub-topics	<b>7</b> 7.1 7.2	<b>Equillibrium</b> Dynamic Equilibrium The Position of Equilibrium	[ <b>5</b> ] 1 4	

Topic	8	Acids and Bases	[5]
Sub-topics	8.1	Theories of Acids and Bases	2
	8.2	Properties of Acids and Bases	1
	8.3	Strong and Weak Acids and Bases	1
	8.4	The pH Scale	1.5
Topic	9	Oxidation and Reduction	[7]
Sub-topics	9.1	Oxidation and Reduction	2
	9.2	Redox Equations	1
	9.2	Reactivity	1
	9.3	Voltaic Cells	1
	9.4	Electrolysis	2
Topic	10	Organic Chemistry	[12]
Sub-topics	10.1	Introduction	4
	10.2	Alkanes	2
	10.3	Alkenes	2
	10.4	Alcohols	1
	10.5	Halogenoalkanes	2
	10.6	Reaction Pathways	1
Topic	11	Measurement and Data Processing	[2]
Sub-topics	11.1	Uncertainty and Error in Experimental Measurement	1
1	11.2	Uncertainties in Calculated Results	0.5
	11.3	Graphical Techniques	0.5
		<u>.</u>	

## Additional Higher Level [55h]

<b>Topic</b> Sub-topics	<b>12</b> 12.1	Atomic Theory Electronic Configuration of Atoms	[ <b>3</b> ] 3
<b>Topic</b> Sub-topics	13 13.1 13.2	Periodicity Periodic Trends Na→ Ar (third period) d-block Elements (first row)	[ <b>4</b> ] 2 2
<b>Topic</b> Sub-topics	14.1 14.2 14.3	Bonding Shapes of Molecules and Ions Hybridization Delocalization of II electrons	[5] 1 2 2
<b>Topic</b> Sub-topics	15.1 15.2 15.3 15.4	Energetic Standard Enthalpy Changes of Reaction Lattice Enthalpy Entropy Spontaneity of Reaction	[ <b>8</b> ] 2 2.5 1.5 2
<b>Topic</b> Sub-topics	16.1 16.2 16.3	Kinetics Rate Expression Reaction Mechanism Activation Energy	[ <b>6</b> ] 3 1 2
<b>Topic</b> Sub-topics	<b>17</b> 17.1 17.2	<b>Equilibrium</b> Liquid-Vapour Equilibrium The Equilibrium Law	[ <b>4</b> ] 2 2
<b>Topic</b> Sub-topics	18.1 18.2 18.3 18.4 18.5	Acids and Bases Calculations involving Acids and Bases Buffer Solutions Salt Hydrolysis Acid-Base Titration Indicators	[ <b>9</b> ] 3.5 1.5 1 2 1
<b>Topic</b> Sub-topics	<b>19</b> 19.1 19.2	Oxidation and Reduction Standard Electrode Potentials Electrolysis	[5] 3 2
<b>Topic</b> Sub-topics	20 20.1 20.2 20.3 20.4 20.5 20.6	Organic Chemistry Introduction Nucleophilic Substitution Reactions Elimination Reactions Condensation Reactions Reaction Pathways Stereoisomerism	[11] 2 2 1 2 1 3

## **Options SL/HL**

NB: Standard level students study the core of these options and higher level student study the whole option (i.e. the core and the extension material)

Option A	<b>Analytical Chemistry</b>	Teaching Hours
Sub-topics	<ul> <li>Core (SL + HL)</li> <li>A.1 Analytical Techniques</li> <li>A.2 Principles of Spectroscopy</li> <li>A.3 Infrared Spectroscopy</li> <li>A.4 Mass Spectrometry</li> <li>A.5 Nuclear Magnetic Resonance Spectroscopy</li> <li>A.6 Atomic Absorption Spectroscopy</li> <li>A.7 Chromatography</li> </ul>	[15] 1 2 3 2 2 3 2 2 3 2
	Extension (HL only)  A.8 Visible and Ultraviolet Spectroscopy  A.9 Nuclear Magnetic Resonance Spectroscopy  A.10 Chromatography	[7] 3 2 2
Option B	Human Biochemistry	Teaching Hours
Sub-topics	Core (SL + HL) B.1 Energy B.2 Proteins B.3 Carbohydrates B.4 Lipids B.5 Vitamins and Minerals B.6 Water B.7 Hormones	[15] 1 3 3 2 2 2 2 2
	Extension (HL only) B.8 Enzymes B.9 Nucleic Acids B.10 Electrolytes	[7] 2.3 3 1.5
<b>Option C</b>	Chemistry in Industry and Technology	Teaching Hours
Sub-topics	<ul> <li>Core (SL + HL)</li> <li>C.1 Iron, Steel and Aluminum</li> <li>C.2 The Oil Industry</li> <li>C.3 Addition Polymers</li> <li>C.4 Catalysts</li> <li>C.5 Fuel Cells and Rechargable Batteries</li> <li>C.6 Liquid Crystals</li> <li>C.7 Nanotechnology</li> </ul>	[15] 3.5 2 2 1.5 2 2 2

	Exten C.8	sion (HL only) Condensations Polymers	[7]
	C.8 C.9	5	2 1
		Silicon and Photovoltaic Cells	
		Liquid Crystals	2 2
Option D	Medio	cines and Drugs	Teaching Hours
Sub-topics	Core (	(SL + HL)	[15]
	D.1	Pharmaceutical Products	2
	D.2		1
	D.3	E	3 3
	D.4	Depressants	
	D.5		2.5
		Antibacterials	2
	D.7	Antivirals	1.5
	Exten	sion (HL only)	[7]
	D.8	Drug Action	2.5
		Drug Design	2.5
	D.10	Mind-Altering Drugs	2
Option E	Envir	conmental Chemistry	Teaching Hours
Sub-topics	Core (	(SL + HL)	[19]
_	E.1	Air	4
	E.2	Water	4
	E.3	Soil	4
	E.4	Waste	3
	Exten	sion (HL only)	[7]
	E.7	Air	2
	E.8	Water and Soil	3
	E.9	Recycling	2
Option F	Food	Chemistry	Teaching Hours
Sub-topics	Core (	(SL + HL)	[15]
-	F.1	Food Groups	2
	F.2	Fats and Oils	3
	F.3	Shelf Life of Foods	4
	F.4	Colour	3
	F.5	Genetically Modified Foods	1
	F.6	Texture	2

	Exten	sion (HL only)	
	F.7	Oxidative Rancidity	1
	F.8	Antioxidants	1
	F.9	Stereochemistry in Foods	2
	F.10	Chemical Structure and Colour	3
Option G	Furth	ner Organic Chemistry	Teaching Hours
Sub-topics	$\underline{\text{Core} (\text{SL} + \text{HL})}$		[15]
	G.1	Electrophilic Addition Reactions	2
	G.2	Nucleophilic Addition Reactions	2
	G.3	Elimination Reactions	2
	G.4	Addition-Elimination Reactions	2
	G.5	Organometallic Compounds	2
	G.6	Reaction Pathways	2
	G.7	Acid-Base Reactions	2
	G.8	Introduction to Arenes	1
	Extension (HL only)		[7]
	G.9	Arenes	2
	G.10	Electrophilic Addition	5