Table of Contents: Lab Work

SCH3UE_07-08

26 Electrolytes: Strong and Weak: Acid –Base: ITC

27 Titration Curve for a strong acid-strong Base

28 Lab: Testing for Polarity of Compounds

29 Lab activity: Polarity of Molecules and IMFA's

30 Planning Lab: testing the structure of Ionic, molecular covalent, Network and metallic substances

31 Enthalpy of Ionic substances: Lattice Energy Vs Hydration Energy

32 Enthalpy of a reaction: Zn + $CuSO_4$

33 Hess's Law: NaOH + HCl