

Table of Contents: SCH 4U Lab Portfolio

SCH4U_08 - 09

| Lab # | Date | Lab Title | Unit | Criteria Assessed |
|-------|------|---|---------------------|-------------------|
| 1 | | How Much? | Moles_Stoichiometry | Planning |
| 2 | | Exo/Endothermic | Thermodynamics | |
| 3 | | Mg + HCl: ΔH | Thermodynamics | |
| 4 | | Zn + Cu ²⁺ : ΔH | Thermodynamics | |
| 5 | | Hess's Law | Thermodynamics | |
| 6 | | C ₂ O ₄ + MnO ₄ | Chemical Kinetics | |
| 7 | | ICT: Maxwell- Boltzmann | Chemical Kinetics | |
| 8 | | Clock Reaction | Chemical Kinetics | |
| 9 | | Order of a Reaction | Chemical Kinetics | |
| 10 | | Heat You Can Eat | Thermodynamics | |
| 11 | | ΔH_C Alcohols | Thermodynamics | |
| 12 | | Equilibrium Simulation (Cheerios/H ₂ O) | Equilibrium | |
| 13 | | Lab Activity: K _c : NO ₂ + N ₂ O ₄ | Equilibrium | |
| 14 | | A Case Study of Haber Process: ICT | Equilibrium | |
| 15 | | Le Chatelier's Principle | Equilibrium | |
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