

## REVIEW: BINARY ACIDS & OXY-ACIDS

SNC2D\_06 -07

1. Name each of the following acids:

HClO <sub>3</sub>	
H <sub>3</sub> PO <sub>4</sub>	
H <sub>2</sub> SO <sub>4</sub>	
HNO <sub>3</sub>	
H <sub>2</sub> CO <sub>3</sub>	
HBrO <sub>3</sub>	
HIO <sub>3</sub>	
HClO <sub>2</sub>	
H <sub>2</sub> SO <sub>3</sub>	
HCl	
HBrO <sub>2</sub>	
HBr	
H <sub>3</sub> PO <sub>3</sub>	
HF	
HBrO <sub>4</sub>	

2. Write the formula for each of the following acids:

Chloric acid	
chlorous acid	
bromic acid	
hydrobromic acid	
nitric acid	
nitrous acid	
phosphoric acid	
phosphorus acid	
hydrochloric acid	
perbromic acid	
hydrofluoric acid	
carbonic acid	
hydroiodic acid	
Hypochlorous acid	