# Lab: Reactions of Aldehydes and Ketones

Reactions of Alkanals and Alkanones

#### Introduction

Ethanal and propanone have been chosen as safe examples of Aldehydes and ketones to illustrate their reactions in simple test-tube experiments. Benzaldehyde has been chosen as an example or an aromatic ketone.

### Materials

ethanal, propanone, benzaldehyde, 2,4-dinitrophenylhydrazine, K<sub>2</sub>Cr<sub>2</sub>O<sub>7 (aq)</sub>, KmnO<sub>4 (aq)</sub> 2M H<sub>2</sub>SO<sub>4</sub>, Hot water-bath, hydroxylamine chloride, cyclohexanone

## **Sa**fetv

Ethanal, propanone and benzaldehyde are irritants to eyes, skin and lungs and are highly flammable. Therefore, you must perform experiment at a fume cupboard, keep the bottles away from flames, keep the stoppers on the bottles as much as possible.

2,4-dinitrophenylhydrazine is toxic (very poisonous), and strongly stains skin to an orange-yellow.

The stain is a cosmetic blemish and disappears within a few days.

- 1. Observe the color and smell of each compound.
- 2. Place a few drops of benzaldehyde on a watch glass and leave in the fume hood. Note any change.
- 3. Into a tt, put 5 drops of ethanal, 2 drops of  $K_2Cr_2O_{7\,(aq)}$  and 10 drops of dilute  $H_2SO_4$ . Shake the tt gently in the hot water-bath. Note your observations. Repeat with propanone and benzaldehyde.
- 4. Repeat procedure 3 using KmnO<sub>4 (aq)</sub> instead of K<sub>2</sub>Cr<sub>2</sub>O<sub>7 (aq)</sub>.
- 5. Add 4cm³ of conc. H<sub>2</sub>SO<sub>4</sub> to 2g of 2,4-dinitrophenylhydrazine, and then carefully add 30cm<sup>3</sup> of methanol. Warm gently on a water bath until a clear solution forms, and add 10cm<sup>3</sup> of distilled water.
  - Add 5cm<sup>3</sup> of the above solution to 1cm<sup>3</sup> of ethanal. Observe any reactions. Repeat using propanone and benzaldehyde instead of ethanal.
- 6. Dissolve 1g of hydroxylamine chloride in 4cm<sup>3</sup> of warm water containing 1.5g of sodium ethanoate crystals. Add 1cm<sup>3</sup> of propanone. Observe any changes Repeat using cyclohexanone instead of propanone.

<u>Procedure</u>