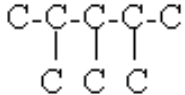
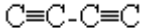
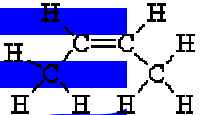
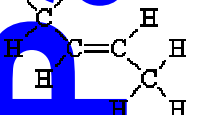
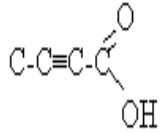
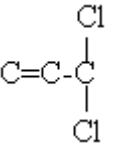
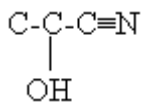
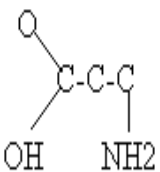
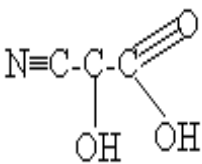
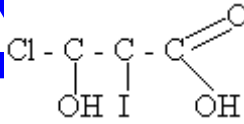

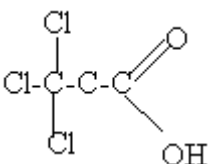
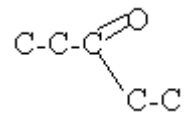
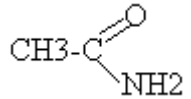


# CARBON CHEMISTRY – NOMENCLATURE CHART

Alkane ( $C_nH_{2n+2}$ )	Alkene ( $C_nH_{2n}$ )	Alkyne ( $C_nH_{2n-2}$ )	Haloalkene ( $RX$ )	Alkanol ( $R-OH$ )	Amines ( $RNH_2$ )	Ethers ( $R-OR$ )
<ul style="list-style-type: none"> <li>Bonded to each other with single bond</li> <li>bond angles of <math>109.5^\circ</math></li> <li><b>SUFFIX: ANE</b></li> </ul>  <p>2,3,4-trimethylpentane</p> <ul style="list-style-type: none"> <li>Non-Polar</li> <li>London Dispersion Forces</li> </ul>	<ul style="list-style-type: none"> <li>Double bond</li> <li><b>SUFFIX: ENE</b></li> </ul>  <ul style="list-style-type: none"> <li>but-1,3-diene</li> </ul> <p><i>cis-2-butene:</i></p>  <p><i>trans-2-butene:</i></p> 	<ul style="list-style-type: none"> <li><math>C_nH_{2n-2}</math></li> <li>triple bond</li> <li><b>SUFFIX: YNE</b></li> </ul>  <ul style="list-style-type: none"> <li>2-yne-butanoic acid</li> <li>Non Polar</li> <li>London Dispersion Forces</li> </ul>	<ul style="list-style-type: none"> <li><math>C_nH_{2n+1}K</math></li> <li>halogenocompounds</li> <li>haloalkene</li> </ul>  <ul style="list-style-type: none"> <li>1,1-dichloropropene</li> <li>Polar</li> <li>Dipole-Dipole</li> </ul>	<ul style="list-style-type: none"> <li><math>(C_nH_{2n+1})OH</math></li> <li>alcohols</li> <li>alkanols</li> <li>prefix: hydroxy</li> <li>suffix: ol</li> <li>classes: 1, 2, 3</li> </ul>  <p>2-Hydroxypropanoic acid</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Hydrogen Bond</li> </ul>	<ul style="list-style-type: none"> <li><math>NH_2</math></li> <li><b>SUFFIX: AMINE</b></li> <li><b>PREFIX: AMINO-</b></li> <li><b>Classes: 1, 2, 3</b></li> </ul>  <p>3-amino-propanoic acid</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Hydrogen Bond</li> </ul>	<ul style="list-style-type: none"> <li>Alk-oxy-alkane</li> </ul> <p>C-O-C-C methoxy ethane</p> <ul style="list-style-type: none"> <li>Non Polar</li> <li>London Dispersion Forces</li> </ul>
<p><b>Nitriles (<math>R-CN</math>)</b></p> <ul style="list-style-type: none"> <li><math>-C\equiv N</math></li> <li><b>SUFFIX: NITRILE</b></li> <li><b>PREFIX: CYNO</b></li> </ul>  <p>3-cyano-2-hydroxypropanoic acid</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Dipole-Dipole</li> </ul>	<p><b>Alkanal (<math>R-CHO</math>)</b></p> <ul style="list-style-type: none"> <li>aldehydes</li> <li>alkanal</li> <li><b>PREFIX: OXO</b></li> </ul>  <p>3-chloro-3-hydroxy-2-iodopropanoic acid</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Dipole-Dipole</li> </ul>	<p><b>Alkanone</b> </p> <ul style="list-style-type: none"> <li><b>PREFIX: OXO</b></li> <li><b>SUFFIX: ONE</b></li> <li>ketones</li> <li>3-pentanone</li> <li>Polar</li> <li>Dipole-Dipole</li> </ul>	<p><b>Alkanoic Acid (<math>R-COOH</math>)</b></p> <ul style="list-style-type: none"> <li>carboxylic acid</li> <li>alkanoic acid</li> </ul>  <p>3,3,3-trichloropropanoic acid</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Dipole-Dipole</li> <li>Hydrogen Bond</li> </ul>	<p><b>Esters (<math>RCOOR'</math>)</b></p>  <p>methyl propanoate</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Dipole-Dipole</li> </ul>	<p><b>Amides (<math>R-CONH_2</math>)</b></p>  <p>ethanamide</p> <ul style="list-style-type: none"> <li>Polar</li> <li>Dipole-Dipole</li> <li>H-bonding</li> </ul>	